

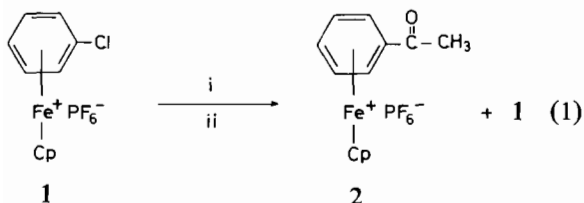
Synthesis and Characterization of η^6 -Acetophenone- η^5 -cyclopentadienyliron Hexafluorophosphate

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The aluminium chloride catalyzed ligand exchange reaction between ferrocene (FcH) and arenes, first described by Nesmeyanov [1] has been extensively used in the preparation of sandwich complexes of the type arene $\text{FeCp}^+\text{PF}_6^-$ [2]. Nonetheless, the need to use aluminium chloride severely limits the possible arene substituents. To the best of our knowledge, no synthesis of CpFe^+ complexed ketoarene via direct ligand exchange between ketoarene and FcH was reported. The present note describes an indirect way of preparing the title compound, $\text{C}_6\text{H}_5\text{COCH}_3\text{-FeCp}^+\text{PF}_6^-$ (2) in good yield (eqn. (1)):



(i) $\text{K}_2\text{CO}_3 / \text{CH}_3\text{CH}_2\text{NO}_2$; (ii) $\text{HCl} / \text{C}_2\text{H}_5\text{OH}$

Activation toward $\text{S}_{\text{N}}\text{Ar}$ substitution of normally unreactive arene halides by complexation to transition metal cations is well known [2, 3]. Recently, we have discovered that the organometallic complex, $\text{C}_6\text{H}_5\text{ClFeCp}^+\text{PF}_6^-$ (1) could readily undergo $\text{S}_{\text{N}}\text{Ar}$

substitution of the chloride atom with enolate anions to give corresponding substituted benzene complexes [4]. We have now applied this method for the synthesis of cation 2. The reaction of the cation 1 with an excess of nitroethane ($\text{C}_2\text{H}_5\text{NO}_2$) in the presence of 3.0 molar equivalent of K_2CO_3 for 30 h at RT under N_2 , followed by HCl /Ethanol work up, gave 1/1 mixture of 1 and 2 [5]*. The η^6 -acetophenone- η^5 -cyclopentadienyliron cation (2) can be obtained in a pure form (35–40% yield) after chromatographic separation through alumina column with elution by acetone. *Anal. Calc.* for $\text{C}_{13}\text{-H}_{13}\text{OFePF}_6$: C, 40.44; H, 3.39. Found: C, 40.33; H, 3.40%. Melting point 171–173 °C. IR (KBr, cm^{-1}) 1740 (CO). $^1\text{H NMR}$ (acetone- d_6) 6.9–7.2 (m, 2H), 6.6–6.9 (m, 3H) aromatic; 5.3 (s, 5H, Cp) and 2.8 (s, 3H, CH_3).

In conclusion, the nucleophilic substitution of the chlorine atom in cation 1 with carbanion derived from nitroethane provided a convenient route to the synthesis of cation 2.

References

- 1 A. N. Nesmeyanov, N. A. Vol'kenau and I. N. Bolesova, *Dokl. Akad. Nauk SSSR*, 149, 615 (1963).
- 2 R. G. Sutherland, *J. Organomet. Chem.*, 3, 311 (1977); D. Astruc, *Tetrahedron*, 39 (24), 4027 (1984).
- 3 M. F. Semmelhack, G. R. Clark, J. L. Garcia, J. J. Harrison, Y. Thebtaranonth, W. Wulff and A. Yamashita, *Tetrahedron*, 37, 3959 (1981).
- 4 R. M. Moriarty and U. S. Gill, *Organometallics* (1985) in press.
- 5 J. March, 'Advanced Organic Chemistry', 2nd edn., McGraw Hill, New York, 1977, p. 808.

*Nef reaction.